



# Improvement of **Odor Assessment** in a **Life Cycle Assessment** Framework

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## Introduction

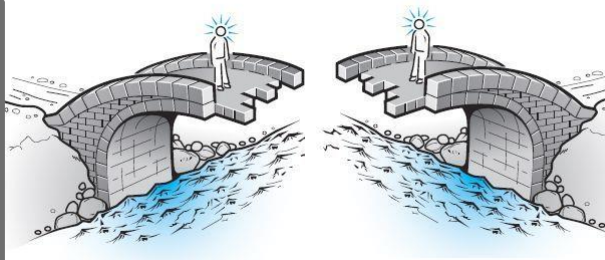
Many food production systems involve components that generate odor

Incidence of odor annoyance is increasing

Odor plays a role in the selection of technologies for manure and waste water management

**AIM: develop simple method for odor midpoint assessment in life cycle assessment**

Life cycle  
impact  
assessment  
methods



Research on  
odor threshold  
values

## History

Heijungs et al. (1992):

- *OTV*: concentration ( $\text{kg}\cdot\text{m}^{-3}$ ) at which a chemical is detectable by 50% of the population
- Malodorous air =  $\sum_i \frac{m_{i,air}}{OTV_{i,air}}$
- Does not consider persistence of odor



Marchand et al. (2013):

- Temporal component added
- USEtox (adjusted) as a basis for fate modeling of odorants
- Ultimate odor burden is expressed in terms of 11 different midpoint indicators

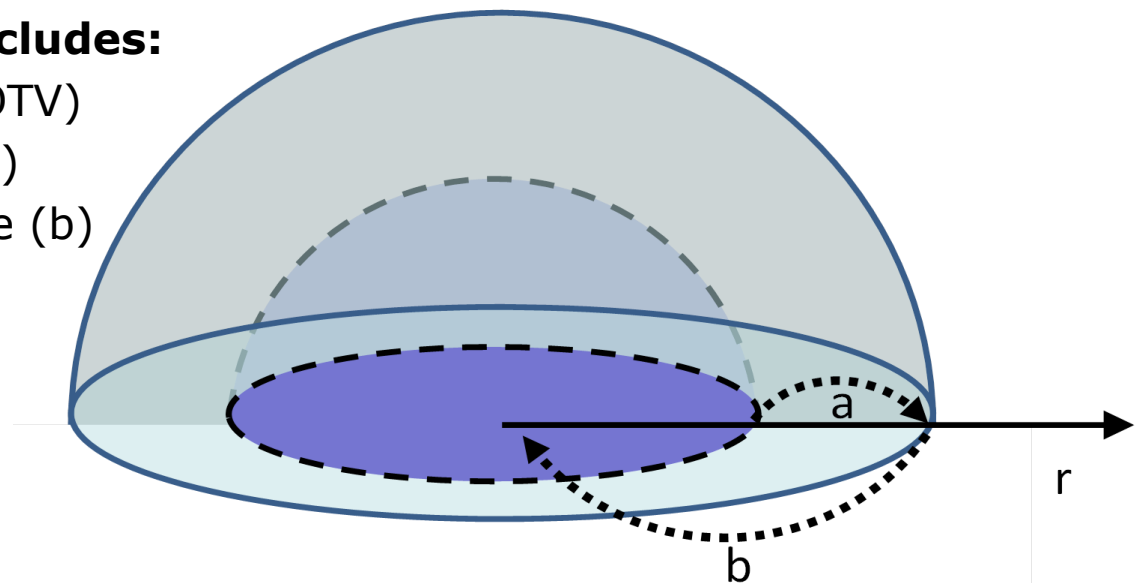
## Methods

### Non-site-specific midpoint method

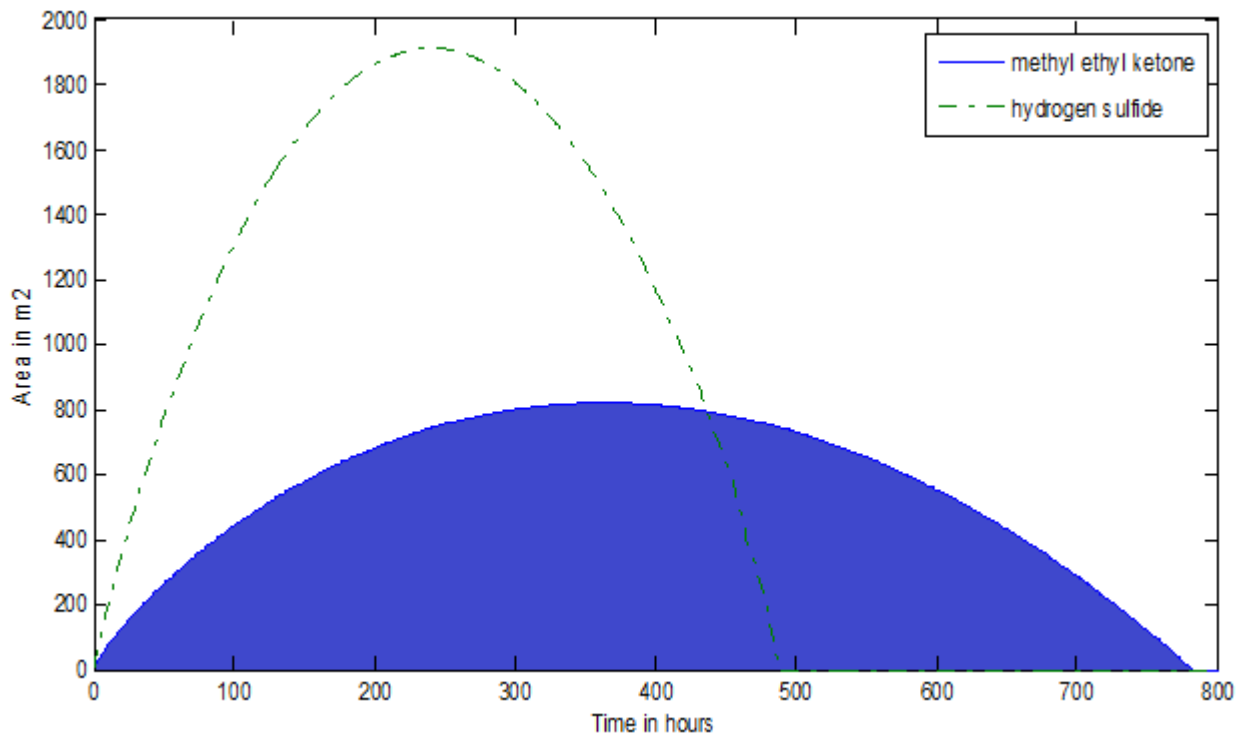
- Wind-induced turbulence not included (equal for all odorants)
- Focus on chemical specific factors causing an odorant's potential impacts
- Detection levels often orders of magnitudes lower than toxicological effects

### Odor footprint includes:

- Odor potency (OTV)
- Diffusion rate (a)
- Degradation rate (b)



## Results



	OTV	Diffusion	Degradation
Hydrogen sulfide	<b>LOW</b>	<b>HIGH</b>	HIGH
Methyl ethyl ketone	HIGH	LOW	<b>LOW</b>

## Results

Characterization factors for 33 key odorants (Peters et al., 2014)

Expression of odor by Heijungs et al. (1992): "malodorous air" for H<sub>2</sub>S is more than 2000 times higher than for butanone

Odor footprint of H<sub>2</sub>S is 46% higher than butanone

Difference between these relationships is due to inclusion of diffusion and degradation in the latter (both high for H<sub>2</sub>S)

Parameter	OTV	D	k <sub>OH</sub>	Odor footprint	
Odorant	g·m <sup>-3</sup>	m <sup>2</sup> ·s	s <sup>-1</sup>	m <sup>2</sup> s	kg H <sub>2</sub> S-eq·kg <sup>-1</sup>
Hydrogen sulfide	<b>5.72x10<sup>-7</sup></b>	2.31x10 <sup>-5</sup>	7.05x10 <sup>-6</sup>	2.30x10 <sup>9</sup>	<b>1.00</b>
Butanone	<b>1.30x10<sup>-3</sup></b>	1.29x10 <sup>-5</sup>	1.80x10 <sup>-6</sup>	1.57x10 <sup>9</sup>	<b>0.68</b>

## Discussion

**Pathway for improvement** of LCA methodology for odorous emissions

**Advantage** compared to...

- ... Heijungs et al.: **temporal aspect** of odorant impact **included**
- ... Marchand et al.: more **simple** implementation

**Persistence** of an odorant is **included**

**Shortcomings:**

- **Synergetic** effects **not included**
- Impacts of **degradation products** **not considered**
- **Other reactants** than hydroxyl radicals are **not included**



## Conclusion

This **approach**...

... is an **improvement** of the **traditional LCA approach** for raking odorants, as not only the **potency** of an odorous emission is expressed, but also its **diffusion** and **degradation**

... is a **midpoint method** which may be **appropriate** for use when **odor impacts** are **considered significant** in a system to be analyzed using **LCA**, but a **site-generic approach** is needed



**Interested?** Read:

Peters G.M., Murphy K.R., Adamsen A.P., Bruun S., Svanström M., ten Hoeve M. (2014). Improving odour assessment in LCA - the odour footprint. Accepted by *Int J Life Cycle Assess*, 4 July 2014